Structural determinants for interaction of partial agonists with acetylcholine binding protein and neuronal α7 nicotinic acetylcholine receptor

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The pentameric acetylcholine-binding protein (AChBP) is a soluble surrogate of the ligand binding domain of nicotinic acetylcholine receptors. Agonists bind within a nest of aromatic side chains contributed by loops C and F on opposing faces of each subunit interface. Crystal structures of Aplysia AChBP bound with the agonist anabaseine, two partial agonists selectively activating the α7 receptor, 3-(2,4-dimethoxybenzylidene)-anabaseine and its 4-hydroxy metabolite, and an indole-containing partial agonist, tropisetron, were solved at 2.7–1.75 Å resolution. All structures identify the Trp147 carbonyl oxygen as the hydrogen bond acceptor for the agonist-protonated nitrogen. In the partial agonist complex, the benzylidene and indole substituent positions, dictated by tight structural and functional surrogate for the extracellular LBD (Lymnaea stagnalis (L-AChBP) and Bulinus truncatus, shows binding affinities and specificities similar to those of the neuronal, α7 nAChR subtype for nicotinic agonists and antagonists, as exemplified by its lower affinity for acetylcholine, but higher affinity for the α7-specific antagonistic α-conotoxin peptides (Hansen et al., 2002, 2004; Celie et al., 2004). The coupling of AChBP with the pore domain of the 5HT1A receptor not only results in acetylcholine binding with modest or intermediate affinity, characteristic of activatable receptors, but also triggers a low frequency opening of the ion channel (Bouzat et al., 2004), arguing for AChBP to be both a structural and functional surrogate for the extracellular LBD of nAChRs. A refined electron microscopy structure of the heteropentameric muscle-type, x1β2γδ nAChR, solved in part

Introduction
Nicotinic acetylcholine receptors (nAChRs) are prototypical cation-selective, ligand-gated ion channels (LGIC) that mediate fast neurotransmission in the central and peripheral nervous systems (Changeux and Edelstein, 2005; Taylor, 2006). The nAChRs belong to the Cys-loop superfamily of LGICs, that in mammals also includes the serotonin, 5HT3; γ-aminobutyric acids, GABA-A and GABA-C; and glycine receptors, and are formed by distinctive combinations of five subunits that confer selectivity in pharmacological properties and cellular location. The diversity of nAChR subunit assembly and subtype is most evident in the central nervous system, in which nine α and three β subunits have been defined.

The discovery of the acetylcholine binding protein (AChBP) from a freshwater snail, followed by its crystallographic analysis, showed a homopentameric assembly of subunits homologous to the N-terminal, extracellular ligand-binding domain (LBD) of the nAChR (Brejc et al, 2001). In addition to the overall primary and tertiary structural similarity of the subunits, the aromatic residues that form the ligand binding pocket at the subunit interface are highly conserved in the nAChR family, although the interface residues that contribute to the complementary face (or (-) face) show greater variability than those on the principal face (or (+) face). The binding pocket of AChBP possesses all the functional residues identified in the nAChR LBD, and extended areas of the subunit interface in the apical and membrane directions provide multiple means for selective accommodation of the many nicotinic ligands (Hansen et al., 2005). AChBP from Aplysia californica (A-AChBP), compared with its relatives from Lymnaea stagnalis (L-AChBP) and Bulinus truncatus, shows binding affinities and specificities similar to those of the neuronal, α7 nAChR subtype for nicotinic agonists and antagonists, as exemplified by its lower affinity for acetylcholine, but higher affinity for the α7-specific antagonistic α-conotoxin peptides (Hansen et al., 2002, 2004; Celie et al., 2004). The coupling of AChBP with the pore domain of the 5HT1A receptor not only results in acetylcholine binding with modest or intermediate affinity, characteristic of activatable receptors, but also triggers a low frequency opening of the ion channel (Bouzat et al., 2004), arguing for AChBP to be both a structural and functional surrogate for the extracellular LBD of nAChRs. A refined electron microscopy structure of the heteropentameric muscle-type, x1β2γδ nAChR, solved in part
using the AChBP template (Unwin, 2005), and the crystal structure of the extracellular domain of the isolated muscle-type α1 subunit bound to the peptide antagonist, α-bungarotoxin (Dellisanti et al., 2007), confirms the close structural similarity between the AChBP and nAChR subunits. A recent characterization of pentameric, prokaryotic LGICs shows their structural homology to AChBP and documents the similarity of their intra-subunit and inter-subunit arrangements (Bocquet et al., 2007, 2009; Hilf and Dutzler, 2008, 2009).

To date, AChBP offers the best template for obtaining high-resolution structures of the LBD of nAChRs. In turn, structural studies of AChBP in complex with a large variety of nAChR agonists and competitive antagonists have shown that loop C, found at the outer perimeter of the pentamer, adopts distinctive conformations upon agonist and antagonist occupation of the binding pocket (Bourne et al., 2005; Hansen et al., 2005), a phenomenon that can also be monitored in solution by hydrogen–deuterium exchange mass spectrometry (Shi et al., 2006). Overall, a ‘core agonist signature motif’ that recognizes the activating ligands was localized central to the binding pocket. In contrast to the small agonist molecules, the larger antagonists occupy an expanded surface area at the subunit interface resulting in further opening of loop C and often conferring a greater selectivity than the agonists do for receptor subtypes.

In comparison with full agonists or antagonists, partial agonists elicit only a fractional pharmacological response, even at full binding site occupation (Stephenson, 1956; Pratt and Taylor, 1990; Hoyer and Borddeke, 1993). Using state functions to describe receptor activation, partial agonism can be explained by the occupied ligand not shifting the conformational equilibrium between open and closed states fully to the open channel state (Pratt and Taylor, 1990). A recent proposal suggests that partial agonism in the nAChR superfamily is associated with a pre-open conformation that has a higher affinity for agonists than the resting receptor (Lape et al., 2008). In contrast to full agonists, partial agonists would have a diminished capacity to occupy the pre-open state before opening the channel. Irrespective of the mechanism and the structural description of the ligand-bound states, a ceiling on agonist efficacy can serve to minimize the toxicity upon overdose and reduce addiction liability of drugs. Achieving receptor subtype selectivity, affinities approaching or exceeding that of nicotine, and partial agonist characteristics for nAChR stimulation are all desirable features sought to improve nicotinic receptor-targeted therapies for neurodegenerative and psychiatric disorders (Kem, 2000; Hogg and Bertrand, 2007).

Recent studies have focused on a series of anabaseine-derived compounds showing a mixed pharmacological profile towards nAChRs (Briggs et al., 1995; de Fiebre et al., 1995; Kem et al., 1997, 2004). The parent molecule, anabaseine (Figure 1), is a natural nicotine-related pyridine alkaloid used by certain marine worms (Phylum Nemertinea, ribbon worms) as a chemical defense against predators and as a means for capturing prey (Kem et al., 2006a). It is a relatively non-selective nAChR agonist, but activates the muscle-type α1β2γδ and neuronal α7βδ subtypes of nAChRs with high potency and full efficacy (Kem et al., 1997). However, addition of a benzylidene group at the 3-position of the anabaseine tetrahydropyridine ring, generating a benzylidene-substituted anabaseine (BA), is sufficient to confer functional selectivity for α7 nAChRs (de Fiebre et al., 1995; Papke et al., 2004). The large number of BA analogues synthesized using various substituents on the benzylidene ring and varying degrees of agonist efficacy provide a series of congeners well suited for a detailed analysis of ligand binding sites of nAChRs and AChBPs. Moreover, the BA analogues show unique absorption and fluorescence emission properties that enable one to describe the protonation state of the bound ligand, and the permittivity and polarizability of the surrounding side chains in the binding site (Talley et al., 2006).

The BA derivative, 3-(2,4-dimethoxybenzylidene)-anabaseine (DMXBA, Figure 1) is also a promising drug candidate, as the added hydrophobic substituents favour the penetration of the blood–brain barrier and confer partial agonist activity at the α7 receptor. DMXBA is in clinical trials for cognition enhancement and improvement of auditory gating in schizophrenia (Olincy et al., 2006; Freedman et al., 2008). It also shows neuroprotective properties (Martin et al., 1994; Shimohama et al., 1998). After oral administration, DMXBA is transformed into three hydroxy metabolites, 2-OH-, 4-OH- and 2,4-di-OH-DMXBA, which show binding affinities and partial agonist properties superior to the parent compound found on rat and human α7 nAChRs (Kem et al., 2004). However, these metabolites are more polar than DMXBA and cross the blood–brain barrier less readily, potentially limiting their use as therapeutic agents targeting the central nervous system. DMXBA and its hydroxy metabolites are low potency antagonists at the human 5HT1A receptor; the metabolites are partial agonists at the murine 5HT3 receptor (Machu et al., 2001; Zhang et al., 2006).

Tropisetron was initially developed as a high affinity antagonist for 5HT3 receptors. More recently, it was identified as a partial agonist selective for the α7 receptor and an antagonist for non-α7 receptors (Macor et al., 2001). In several countries, it is used to alleviate chemotherapy-induced nausea and vomiting (Ho and Gan, 2006). Tropisetron contains bicyclic tropane and hydrophobic indole moieties, in which the bridged azo-nitrogen and indole ester components may adopt similar positions, respectively, to the imine-nitrogen and pyridine ring components of the anabaseines (Figure 1). Moreover, various tropine esters
show pharmacological activities similar to the BA α7-selective partial agonists.

To understand the structural determinants that confer partial agonist character and dictate nAChR subtype selectivity, we have carried out a comprehensive structural study of α7-selective partial agonists using A-ChBP as a surrogate of the extracellular LBD of the α7 nAChR. The crystal structures of A-ChBP bound with the non-selective full agonist anabaseine, two α7-selective benzylidene-anabaseine derivatives, DMXBA and 4-OH-DMXBA, and the α7-selective, though chemically-distinct, tropisetron, were solved to 2.7–1.75 Å resolution range. These structures show at least two modes of binding of partial agonists and identify essential interactions contributing to the high affinity binding of these compounds to the α7 nAChR.

Results and discussion

Overall views of the structures

The structures of the A-ChBP complexes with anabaseine, the BA derivatives DMXBA and 4-OH-DMXBA, and tropisetron (Figure 1), were solved from crystals grown in three chemically-distinct, tropisetron, were solved to 2.7–1.75 Å resolution range. These structures show at least two modes of binding of partial agonists and identify essential interactions contributing to the high affinity binding of these compounds to the α7 nAChR.

Table 1 Data collection and refinement statistics

<table>
<thead>
<tr>
<th>Data collection</th>
<th>Anabaseine</th>
<th>DMXBA</th>
<th>4-OH-DMXBA</th>
<th>Tropisetron</th>
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<td>ID23-EH1</td>
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<td>25.1 (4.2)</td>
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<td>99.9 (99.9)</td>
<td>99.7 (99.5)</td>
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<td></td>
<td>Rwork/Rfree (%)</td>
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<td>1.10</td>
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<td></td>
<td>Bond angles (deg)</td>
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<td>1.507</td>
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<td>2WNJ</td>
<td>2WN9</td>
<td>2WNC</td>
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* Values in parentheses are those for the last shell.
* Rmerge = \( \frac{\sum |I_{hkl}|-\left\langle |I_{hkl}|\right\rangle}{\sum |I_{hkl}|} \), where \( I \) is an individual reflection measurement and \( \left\langle I \right\rangle \) is the mean intensity for symmetry-related reflections.
* Rmerge = \( \frac{\sum |I_{hkl}|/|F_{o}| - |F_{c}|/|F_{o}|}{\sum |I_{hkl}|} \), where Fo and Fc are observed and calculated structure factors, respectively. Rmerge is calculated for 3% of randomly selected reflections excluded from refinement, except for DMXBA transferred from the 4-OH-DMXBA dataset.
* t.m.s.d. from ideal values.
Instead of retaining the predicted co-planar conformation (Supplementary Table I), the two rings (tetrahydropyridine and pyridine) are twisted by ~35° and ~45° relative to each other, as found for the bound nicotine (Celie et al., 2004). In conformer A, the imine nitrogen is properly positioned to form a hydrogen bond (~2.7 Å) with the carbonyl of Trp 147 on the (+) face of the interface (Figure 3). The tetrahydropyridine ring in anabaseine is bound in a position similar to the pyrrolidine ring in nicotine (Celie et al., 2004) and the bridged nitrogen in epibatidine (Hansen et al., 2005). The tetrahydropyridine ring in anabaseine which, compared with the pyrrolidine ring of nicotine, is larger and tilted by ~80° towards loop C, adopts a flattened boat conformation. The ring establishes near face-to-edge stacking interactions with Trp 147 on the (+) face and Tyr 188 in loop C, and face-to-face interactions with Tyr 195 at the hinge point of loop C and the Cys 190–Cys 191 disulfide at its tip. In conformer B, the rotated tetrahydropyridine ring loses its hydrogen bond with Trp 147 but maintains an optimal interaction with the nest of aromatic residues. The pyridine nitrogen is hydrogen bonded, through a bridging water molecule, to the main chains of residues Ile 106 and Ile 118, as previously observed for nicotine (Celie et al., 2004) and the neonicotinoids: imidacloprid and thiacloprid (Talley et al., 2008). Residues Val 108, Met 116, Ile 118 and Val 148 also form van der Waals contacts with the anabaseine pyridine ring. In contrast to its contribution in the nicotine and epibatidine complexes, Tyr 93 at the base of the binding pocket is shifted away (by ~1.5 Å) from anabaseine and only weakly contributes to its binding (Figure 3).

The mode of binding of the cyclic form of anabaseine, wherein aromatic residues from loop C contribute to stabilization of the tetrahydropyridine ring, is similar to those of the other full agonists, nicotine and epibatidine (Figure 3). In turn, loop C adopts a conformation similar to that seen in complexes with full agonists with the disulfide that abuts at the midpoint between the two rings (Celie et al., 2004; Hansen et al., 2005). In the tetrahydropyridine ring of anabaseine, the absence of a protruding N-methyl group pointing toward loop C, as found in nicotine or the nitrogen-containing alicyclic skeleton (bridge ring) in epibatidine, could be responsible for the lower affinity of anabaseine for A-AChBP (Table II). The lack of tight interaction between anabaseine and Tyr 55 on the (+) face may contribute to the relatively low affinity of anabaseine for A-AChBP. In L-AChBP and rat brain a7, the bulky Trp side chain that replaces A-AChBP Tyr 55 on the (+) face most probably contributes to the respective 4-fold and 17-fold higher affinity (Table II), consistent with additional stacking interactions with the anabaseine tetrahydropyridine ring (Figure 3).

In the third binding site, the open chain ammonium ketone of anabaseine is present with an open conformation of loop C, whereas in the fourth binding site, the closed form overlaps with the open chain form associated with a non-tethered loop C (Figure 2A). The mode of binding of the open form of anabaseine (conformer C) largely differs from that of the closed form by a flip of the molecule such that the pyridine ring is shifted 7 Å outward and the ammonium group lies within the binding pocket. In one binding site, the open chain form is bound in the opposite direction, with the pyridine ring in stacking interaction with Trp 147, whereas the ammonium group points toward the solvent. Only the cyclic form

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**The anabaseine complex**

Anabaseine is a full agonist at muscle-type and α7 nAChRs (Figure 1; Table II). Similar to nicotine and its analogues, anabaseine is composed of a non-aromatic tetrahydropyridine ring attached to the 3-position of an aromatic pyridine ring. However, in the anabaseine tetrahydropyridine ring, the imine double bond is electronically conjugated to the aromatic ring, instead of being an isolated tertiary or secondary amine as in nicotine or epibatidine.

In two out of the five binding sites in the pentamer, the cyclic form of anabaseine adopts two distinct conformations, A and B, that differ by a 90° rotation of the tetrahydropyridine ring relative to the pyridine ring (Figure 2A). The anabaseine complex

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**Figure 2** The four pentameric full and partial agonist–AChBP complexes: overall views. (A) Anabaseine (orange bonds), (B) DMXBA (bright orange and pink), (C) 4-OH-DMXBA (light orange) and (D) tropisetron (green) are shown as bound at each subunit interface. In the first column, the four A-AChBP pentamers are viewed from their ‘membrane’ side. The conformational flexibility of the tip of the loop C is clearly visible. In the second column, the pentamers are oriented with their apical N-terminus at top and their ‘membrane’ side C-terminus at bottom. The bound ligands are shown through a transparent surface. Third column shows close-up views of the bound ligands with their 2.7-1.75 Å resolution 2Fo–Fc electron density maps contoured at 1.0σ (blue). The subunits contributing the (+) and (−) faces of one interface are shown in yellow and cyan, respectively. The ligands are shown with red oxygen and blue nitrogen atoms.
Table II Ligand dissociation constants for A- and L-AChBP and other Cys loop receptors

<table>
<thead>
<tr>
<th>Ligand</th>
<th>A-AChBP (Kd, nM)</th>
<th>L-AChBP (Kd, nM)</th>
<th>Rat α7</th>
<th>Rat α4β2</th>
<th>Rat 5HT3</th>
<th>Human 5HT3 (IC50, µM)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Anabaseine</td>
<td>&gt;1000</td>
<td>240 ± 21</td>
<td>200 ± 24</td>
<td>110 ± 14</td>
<td>—</td>
<td>—</td>
</tr>
<tr>
<td>DMXBA</td>
<td>330 ± 58</td>
<td>19 ± 1</td>
<td>130 ± 14</td>
<td>253 ± 37</td>
<td>—</td>
<td>3.1±</td>
</tr>
<tr>
<td>4-OH-DMXBA</td>
<td>3.0 ± 0.96</td>
<td>0.43 ± 0.03</td>
<td>235 ± 14</td>
<td>69 ± 30</td>
<td>—</td>
<td>7.4±</td>
</tr>
<tr>
<td>2-OH-DMXBA</td>
<td>220 ± 26</td>
<td>12 ± 3</td>
<td>317 ± 67</td>
<td>387 ± 25</td>
<td>—</td>
<td>1.5±</td>
</tr>
<tr>
<td>Tropisetron</td>
<td>479 ± 57</td>
<td>74 ± 6</td>
<td>6.9 ± 2.4</td>
<td>55 000 ± 28 000</td>
<td>5.3 ± 3.0</td>
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</tbody>
</table>

*Counts were measured by the displacement of the agonist [3H] epibatidine.

**Counts were measured by the displacement of the partial agonist [3H]-cytisine.

***Counts were measured by the displacement of the antagonist 125I-bungarotoxin; [α4β2 Kd value measured by the displacement of the partial agonist [3H]-cytisine.

****Counts were measured by the displacement of the antagonist 125I-bungarotoxin; [α4β2 Kd value measured by the displacement of the partial agonist [3H]-cytisine.

This study [AChBP Kd values measured by displacement of the agonist [3H] epibatidine.

Macor et al., 2001 [α4β2 Kd value measured by displacement of the partial agonist [3H]-cytisine.

binds with significant affinity to the ACh binding site of muscle and neuronal nAChRs (Kem et al., 2006b). The lack of full site occupancy of the pentameric AChBP by anabaseine relates to the presence of comparable amounts of the cyclic and ring opened ammonium ketone forms at the pH used for crystallization.

**The DMXBA complex**

The structure of A-AChBP in complex with DMXBA shows an agonist molecule tightly associated with each of the five subunit interfaces (Figures 1 and 2B). As in the anabaseine complex, a similar network of hydrogen bonds is involved in ligand binding, with the protonated imine nitrogen ideally positioned to target the Trp 147 carbonyl oxygen, ~2.75 Å away, whereas the pyridine amine is bound to Ile 106 and Ile 118 main chains through a water molecule (Figure 4). As expected from the bathochromic shifts in the bound DMXBA–AChBP complex (Talley et al., 2006), aromatic interactions are also conserved with near parallel stacking of the pyridine ring with the Trp 147 indole ring, whereas the tetrahydropyridine ring makes close contact with the Tyr 93 phenolic ring on the membrane side of the (−) face. The benzylidene ring, which confers α7 selectivity, protrudes from the binding pocket in a direction parallel to the radial axis of the AChBP pentamer. The benzylidene ring readily extends the network of interactions beyond the anabaseine core by interacting with residues from the (−) and (−) faces outside the binding pocket. In comparison with its position in the anabaseine complex, in the DMXBA complex the anabaseine core is slightly displaced (by 0.5 Å) and tilted (by ~15°) towards the ‘membrane’ side of the binding pocket.

In three out of the five binding sites in the pentamer, DMXBA adopts two alternate orientations, A and B, present in roughly equal populations (Figure 2B, Supplementary Table I). The two DMXBA orientations differ by a translational (1.5 Å) and rotational (~25°) drift motion of the anabaseine core which is associated with a ~50° arc motion of the benzylidene ring, resulting in a ~90° arc motion of the Tyr 55 phenol ring and a ~50° rotation of the Tyr 93 side chain (Figure 4). In comparison with the anabaseine complex, the two alternative orientations of the anabaseine core in DMXBA are positioned on both sides of the position occupied by anabaseine. In turn, a slightly weaker hydrogen
bond (3.0 versus 2.7 Å) is predicted between the imine nitrogen and Trp 147 carbonyl in orientation B of the bound DMXBA compared with orientation A. In orientation A, the benzylidene ring is sandwiched between Tyr 188 in loop C on the (+) face and Tyr 55 on the (−) face and projects the distal 4-methoxy group towards a polar side chain triad of Asp 164, Ser 166 and Ser 167 in loop F and close to Thr 36 (3.5 Å) in strand β1 on the (−) face (Figure 4). The 2-methoxy group points in an apical direction to interact with Thr 36, Gln 57 and Ile 118. In orientation B, the rotated benzylidene ring abuts against Cys 190 and is sandwiched between the tip of loop C on the (+) face and Ile 118 on the (−) face. In turn, the 4-methoxy and 2-methoxy groups point towards the solvent and weakly interact with the side chains of Met 116 and Gln 57, respectively. The benzylidene ring of DMXBA points in a direction roughly parallel to the axis of the bulky lycoctonine skeleton of the antagonist methyllycaconitine (MLA) (Figure 4E). In the other two binding sites in the pentamer, the benzylidene ring adopts orientation A favouring interaction with loop F.

Although DMXBA adopts two distinct positional orientations within the binding pocket, the same loop C position is retained (Figures 2B and 4). In fact, the solvent-exposed benzylidene ring in the two orientations prevents loop C from adopting the closed conformation seen for the smaller full agonists, nicotine, epibatidine and anabaseine. Instead, the loop C conformational position is an intermediate between those observed for the full agonists and for ligand-free A-AChBP, respectively (Celie et al., 2004; Hansen et al., 2005).

The 4-OH-DMXBA complex

The structure of A-AChBP in complex with the 4-hydroxy metabolite of DMXBA, 4-OH-DMXBA (Figures 1 and 2C), shows a ligand molecule tightly bound at each subunit
interface, with a position and orientation similar to those seen for conformation A of DMXBA (Supplementary Table I). In turn, a similar network of hydrogen bonds and aromatic interactions is observed for the anabaseine core and the benzylidene ring in the two complexes (Figure 4). However, in contrast to the DMXBA complex, loop C in the 4-OH-DMXBA complex adopts two distinct conformations. In two of the five binding sites, loop C tightly packs against a buried benzylidene ring, with Cys190 making contact with the ring system, and it is displaced outwards by ~2.5 Å relative to its position in structures of AChBP bound to full agonists. In two other binding sites, the tip of loop C does not interact with the benzylidene ring but instead moves outward, by ~7 Å, to adopt an open conformation reminiscent of that seen in ligand-free, anabaseine-bound and DMXBA-bound AChBPs. In the fifth binding site, the tip of loop C is sufficiently disordered to not be visible. Overall, the extent of the loop C opening-to-closure movement on 4-OH-DMXBA binding is up to ~5.5 Å, and this fluctuation is associated with a 0.5 Å outward/radial displacement of the entire 4-OH-DMXBA molecule (Figure 4). However, this movement barely affects (~0.2 Å) the hydrogen bond distance between the imine nitrogen and the Trp147 carbonyl.

Binding experiments have established that 4-OH-DMXBA, with its nanomolar $K_d$ (Table II), shows the highest affinity among a set of related anabaseines for A-AChBP (Talley et al., 2006). Accordingly, the structural data highlight a more favourable polar environment of loop F to accommodate the 4-hydroxyl group of the benzylidene ring. Indeed, substitution of a methyl group to create the 4-methoxy group in DMXBA creates steric clashes between the methyl and the polar environment of loop F to generate a less favourable, alternate conformation B; consistent with a 110-fold decrease in affinity of DMXBA compared with the more potent 4-OH-DMXBA metabolite (Talley et al., 2006). In contrast, the presence of a hydroxyl or a methyl group at the 2-position only moderately affects binding affinity, consistent with greater solvent exposure at this position. In turn, we speculate that 2-OH-DMXBA, when bound, adopts a position similar to that of DMXBA. The importance of having a hydrogen bond-forming moiety at the 4-position is emphasized in the binding data wherein the four BA congeners with the highest affinity for A-AChBP have either a hydroxyl or an amino substitution at the 4-position of the benzylidene ring (Talley et al., 2006). Moreover, our difference spectroscopic analysis shows that in bound 4-OH-DMXBA, both the tetrahydropyridine nitrogen and the phenolic oxygen are protonated. The red shifted peak at 550 nm, characteristic of the zwitterion containing a cationic amine and phenolate anion as a dominant species in solution at pH 7.5, is not evident in the bound species (Talley et al., 2006).

The tropisetron complex

The structure of A-AChBP in complex with tropisetron shows a tightly bound ligand in all five binding sites, with the main axis of the ligand nearly perpendicular to the pentameric five-fold symmetry axis (Figures 1 and 2D). The tropane-bridged ring, with its piperidine moiety in the chair conformation, is positioned similar to the smaller pyrrolidine ring of nicotine or the nitrogen-containing bicyclic skeleton of epibatidine bound to AChBP (Celie et al., 2004; Hansen et al., 2005) (Figure 5). However, the tertiary amine of the tropine establishes a slightly longer hydrogen bonding distance (3.1 Å) from the Trp147 carbonyl than seen for nicotine and epibatidine. The N-methyl group in the tropane ring is lodged between Tyr93, Trp147, Tyr188, Tyr195 from the (+) face and Tyr55 from the (−) face within the aromatic nest. The 100-fold lower $K_d$ compared to A-AChBP (Celie et al., 2004) and (right) epibatidine bound to A-AChBP (Hansen et al., 2005). (C) Superimposition of tropisetron and DMXBA (orientation B) bound to A-AChBP; their benzylidene and indole rings occupy similar positions at the subunit interface.

Figure 5 The tropisetron–AChBP complex: close-up view and structural comparisons. (A) The A-AChBP subunit interface in the tropisetron complex (same orientation as in Figure 2, column 2). (B) Superimposition of tropisetron bound to A-AChBP with (left) nicotine bound to L-AChBP (Celie et al., 2004) and (right) epibatidine bound to A-AChBP (Hansen et al., 2005). (C) Superimposition of tropisetron and DMXBA (orientation B) bound to A-AChBP; their benzylidene and indole rings occupy similar positions at the subunit interface.
indole group relative to the (−) side of the binding pocket is similar to the position of the benzylidene ring in conformation B of bound DMXBA. Consequently, the position and closed conformation of loop C are similar in these two structures: in the tropisetron complex the tip of loop C is only slightly displaced outwards (~1 Å) from its position in the DMXBA complex.

**The pH dependence of ligand binding**

Several types of data indicate that the receptor-binding form of most nAChR ligands, including the anabaseines, is the cationic species. The previously reported absorption difference spectra for the BA derivatives show that the bound species is found in a region of low permittivity and high polarizability, reflecting the abundance of aromatic side chains surrounding the bound ligand (Talley et al., 2006). Moreover, the ionization profiles show the imine hydrogen to be protonated in anabaseine, DMXBA and 4-OH-DMXBA, establishing that the necessary proton is present to establish hydrogen bonding between the imine and the carbonyl oxygen of Trp 147 (Figure 6). Indeed, the imine nitrogen in the anabaseine core in the three compounds progressively loses its proton (and the affinity for the compound decreases) as pH increases from 6.0 to 9.0 (Kem et al., 2004; Talley et al., 2006). The benzylidene anabaseines do not show ring opening, due to π-electron resonance stabilization of the imine by the benzylidene moiety. However, in the case of 4-OH-DMXBA, an additional ionization state (deprotonation of the phenolic hydroxyl) is present. However, it seems from the shift in binding at pH 9.0 that the bound species retains the phenolic hydrogen rather than existing as a zwitterion with the iminium and phenolate. This further establishes the importance of hydrogen bonding through the donor phenol in the bound state of the complex. In contrast to the anabaseines, tropisetron does not show an appreciable pH dependence of binding over the range of 6.0–9.0 (Figure 6). Tropine esters are strong bases with pKa values between 9.8 and 10.0. As such, the bound form should be the protonated species, which is present in appreciable abundance between pH 6.0 and 9.0.

**Partial versus full agonists**

Our study using non-selective and α7-selective agonists highlights several features that shed light on the behaviour of receptor/LBD conformations associated with the binding of partial agonists. First, our structural studies show that ligands with partial agonist characteristics adopt multiple conformations in the bound state (Figure 7). Second, a slight increase in the hydrogen bond distance between the secondary and tertiary amines (the iminium nitrogen is formally a strained tertiary amine) and the backbone carbonyl oxygen on Trp 147, a conserved residue on the (+) face of the binding site, is a conserved feature amongst these ligands. Finally, the loop C position associated with partial agonist binding is not only intermediate between the distinctive positions for agonists and antagonists but also varies between binding sites on the same homomeric pentamer (Figure 7). This again suggests that loop C undergoes rapid opening and closing events around a vacant binding site (Bourne et al., 2005; Shi et al., 2006). In turn, occupation by full versus partial agonists may result in different ligand orientations that are coupled to particular conformations of loop C. The DMXBA- and 4-OH-DMXBA-AChBP structures also indicate that a ligand serving as a partial agonist may adopt a binding pose or configuration at one site distinct from that of a second site within the same pentameric receptor. Indeed, one of the two orientations of

Figure 6 The pH dependence of the binding of the four agonists to AChBP. Competition between the binding of (A) anabaseine, (B) DMXBA, (C) 4-OH-DMXBA and (D) tropisetron with that of [3H]–epibatidine (pKa = 10.1) to α7-AChBP at various pH values, using 0.1 M phosphate/pyrophosphate buffered at pH 6 (■), 7 (▲), 8 (▼) and 9 (◆). The pH dependence of the binding of anabaseine, as well as of the two BAs (Talley et al., 2006), is consistent with the protonated imine (pKa = 7.6) being the bound species. In contrast, the absence of a detectable pH dependence for tropisetron binding in this pH range is consistent with the cationic character of the tropine ester (pKa = 9.8–10.0).
the weak partial agonist DMXBA resembles that of the MLA antagonist, whereas the single orientation of the much more efficacious 4-OH-DMXBA resembles that for agonists (such as lobeline). In other words, orientation A could be that of an agonist, whereas orientation B would be closer to that of an antagonist. A multiplicity of bound nAChR states for partial agonists provides another mechanism for achieving intermediate efficacies for partial agonists. Distinct conformations of congeneric competitive antagonists are found at the ligand binding pocket of AChBP (Gao et al., 2003). Our study is the first to show that partial agonists may also display multiple orientations within the five separate sites in a homomeric pentamer.

Although the soluble AChBP faithfully reflects the recognition properties of nAChRs for nicotinic ligands extending across the range of agonists and antagonists, it probably lacks the capacity to attain all of the conformational states of a functioning receptor tethered to an intrinsic membrane channel. The observation that AChBP fails to show cooperativity upon sequential occupation of its sites by agonist reflects the case in point (Hansen et al., 2002). Despite significant variations in chemical structure, the BAs and tropisetron contain substituted ring systems extending from a hydrogen bond donor of a protonated nitrogen in the imine or tropine. A second common feature of these partial agonists resides in the size of the substituents and their radial orientation when bound, extending their interaction surface outside the binding pocket to a region near loop C on the (−) face. In turn, the substituents control the degree of loop closure and prevent loop C from wrapping around the bound ligand as occurs for full agonists (Figure 7) (Celie et al., 2004; Hansen et al., 2005). Instead, loop C undergoes only limited opening and closure movements and adopts, throughout the five binding sites of a same pentamer, a range of positions as yet uniquely observed for this class of ligands. Recent findings, suggesting that partial and full agonists may interact differently with the binding site that undergoes conformational changes attendant on ligand binding (Lape et al., 2008), are consistent with our structural observations.

**Ligand selectivity for nAChR subtypes**

Anabaseine presents a standard pharmacophore structure, similar to that of nicotine, allowing it to activate α7, muscle and other nAChR subtypes. The addition of the benzylidene group is responsible for the loss of agonist activity at subtypes other than α7. The activity profile of tropisetron is similar to those of the BA α7-selective partial agonists, such as DMXBA or 4-OH-DMXBA. Although tropane and some related agonists containing an additional nitrogen bridging ring (e.g. epibatidine and TC-1698) show non-α7 agonist activity, the tropane-conjugated indole in tropisetron precludes the activation of subtypes other than α7. The sequence alignment of different subunits of the nAChR family suggests that, amongst the loop regions that contribute to the shape of the binding pocket, loop F is a preferred candidate for conferring subtype selectivity to functional regions in the receptors (Supplementary Figure 1). In contrast to loop C, residues in loop F arise from the complementary subunit and show substantial variability in sequence among the nAChRs. Although anabaseine is a full agonist for both the human and rat α7 receptors, DMXBA and its hydroxy metabolites differ in their efficacy for these two receptors (Kem et al., 2004). This discrimination indicates specific interactions of the benzylidene substituents with the receptor. Our structural analysis points to a set of conserved residues in loop F, but not loop C, that determine the relative potency and selectivity of these ligands for the α7 receptor. This is supported by the fact that all BAs produce solvent protection of backbone amide protons in loop F, as shown by hydrogen exchange mass spectrometry (J Shi et al., unpublished results). In electrophysiological studies of chimeric and point mutant α7 receptors, residues in loops C, E and F of the receptor

![Figure 7](https://example.com/f7.png)

**Figure 7** Modes of binding of the nicotinic ligands. (A) Overlap view of the superimposed bound ligands. (B) Schematic representation of the binding modes of a nicotinic full agonist (left), partial agonist (centre) and antagonist (right) to AChBP. The (++) and (−−) faces of one subunit interface are symbolized along with loop C, whose positional conformation varies on binding of the various nicotinic ligands.
LBD that differ across species have been shown to account for the differential pharmacology (Stokes et al., 2004). In particular, our structural data point to a Ser substitution of Gly 166 in loop F of human z7 compared with rat z7, which could contribute to a higher efficacy and potency of the 4-OH-DMXBA metabolite for rat versus human z7 receptors, compared with DMXBA. Ser 166, in addition to neighbouring Asp 163 and Ser 165, provides a more favourable polar environment to accommodate the hydroxyl group at 4-position. Similarly, the position and conformation of tropisetron at the binding interface are consistent with an equal efficacy for the human and rat z7 nAChRs (Stokes et al., 2004).

Conversely, limited modification of a nicotinic ligand, such as the addition of a methyl group to the indole nitrogen of LX278 584, a 5HT2 antagonist structurally related to tropisetron (Barnes et al., 1992), may generate steric clashes with residues in loop F, consistent with a loss of activity on z7 and z4p2 nAChRs (Macor et al., 2001). Hence, loop F represents a major determinant of subtype selectivity among nAChR ligands. Further investigation of other partial agonists with AChBP and how they interact with loop F may provide a more precise understanding of partial agonism in nAChRs.

In summary, our comprehensive structural analysis of AChBP complexes with a non-selective, full nicotinic agonist and three z7-selective partial agonists shows interactions with residue positions in loop F that govern much of the selectivity for these compounds, whereas the closure of loop C is a determinant of agonist efficacy. As the locus of interacting residues within loop F shows high sequence variability within the nAChRs, this region provides a variable surface that should be considered as a template for the design of new subtype-selective drugs with specific pharmacological properties. Further investigation should address the capability of other partial agonists to interact with loop F and induce a variable degree of loop C closure within the binding pocket of nAChRs, and how this might affect the gating process.

In addition, we have shown that this family of partial agonists adopts, at least, two orientations within a given pentameric AChBP molecule. This raises the possibility that partial agonism, in at least some cases, might be due to the degree to which active agonist orientations are adopted within a pentameric nAChR. The influence of multiple bound agonist orientations on other z7 receptor properties, such as cooperativity and desensitization (Papke et al., 2009), may be relevant in understanding the partial agonism for this and related LGIC receptors.

Materials and methods

Nicotinic ligands

Anabaseine and its DMXBA and 4-OH-DMXBA derivatives were synthesized as dihydrochloride salts as described by Kem et al. (2004). Tropisetron hydrochloride and methyllycaconitine citrate (2004). Dulbecco’s modified Eagle’s medium (MediaTech CellGro) was collected every 1–3 days for up to 4 weeks, supplemented with 0.02% NaN3 and stored at 4°C. AChBP was purified on immobilized anti-FLAG M2 affinity gel (Sigma) (Hansen et al., 2002) by elution using 100 μg/ml FLAG peptide in 50 mM Tris–HCl (pH 7.4), 150 mM NaCl, 0.02% NaN3. Pure AChBP was dialysed against the same buffer and concentrated by ultrafiltration. The assembly, as a stable pentamer, was assessed by analytical size-exclusion chromatography.

Radioisogand competition assay

Dissociation constants (Kd) for anabaseine, its derivatives and tropisetron were determined by equilibrium competition with [3H]-epibatidine (Talley et al., 2006). Briefly, AChBP (500 pM in binding sites), polyvinyltoluene (PVT) beads (0.1 mg/ml, Amersham Biosciences), monoclonal anti-FLAG M2 antibody from mouse, and [3H]-epibatidine (20 nM for A-AChBP, 5 nM for L-AChBP) were combined with increasing concentrations of the competing ligand in 0.1 M NaPO4 buffer, pH 7.0 (200 μl total volume). Total and non-specific binding were, respectively, determined in the absence and presence of a saturating concentration (15 μM) of the nicotinic antagonist, MLA. The mixtures were allowed to equilibrate at room temperature for at least 2 h before analysis on a LS 6500 liquid scintillation counter (Beckman Scientific). After background subtraction, data were fitted to a sigmoidal dose–response curve (variable slope) and Kd values were calculated with the observed Ed50 value (Ed50/EC50, 1973) using GraphPad Prism version 4 (San Diego, CA). The reported Kd values (Table II) are represented as means ± s.d. from at least three independent experiments performed in duplicate.

Crystalization and data collection

Crystals were formed with A-AChBP 6.5 mg/ml (260 μM in binding sites; anabaseine complex) or 10.0–12.5 mg/ml (400–500 μM; other three complexes), a 1.5–3-fold molar excess of ligand and, at least, 1 h incubation at room temperature. The crystallization was achieved by vapour diffusion at 18–20°C using a protein-to-well solution ratio of 1:1 (v/v) in 1–2 μl hanging drops. The well solutions were: for the anabaseine complex: 13% PEG-4000, 0.1 M HEPES (pH 7.5), 0.2 M MgCl2; for the DMXBA complex, 13% PEG-4000, 0.1 M HEPES (pH 7.5), 15% isopropanol, 15% glycerol; for the 4-OH-DMXBA complex, 26% PEG-400, 0.1 M HEPES, (pH 7.5), 0.2 M MgCl2; and for the tropisetron complex, 22% PEG-4000, 0.1 M Tris–HCl (pH 7.5), 0.2 M LiSO4. Previous attempts to crystallize an anabaseine complex yielded high-resolution structures, but low binding site occupancy by the ligand. To compensate for the low affinity of anabaseine for A-AChBP (cf. Table II), crystals of the anabaseine complex were further soaked into 20 μl of the well solution supplemented with 0.1 mM of freshly dissolved anabaseine and 20% glycerol (24 h, 18°C). Crystals were flash-cooled in liquid nitrogen, directly (anabaseine, DMXBA, 4-OH-DMXBA complexes) or after a rapid soak in the well solution supplemented with 5% glycerol (tropisetron complex). Data were processed using HKL2000 (Otwinowski and Minor, 1997) or Mosflm (Leslie, 1992). All further computing was carried out with the CCP4 program suite (CCP4, 1994) unless otherwise stated.

Structure determination and refinement

The structures of the four complexes were solved by molecular replacement with AMoRe (Navaza, 1994), using the apo A-AChBP pentamer structure (accession code 2BYN) as a search model. For each complex, the initial model was improved by manual adjustment using Xtalview v4.1 (McRee, 1999) or Coot (Emsley and Cowtan, 2004). The initial models were then refined with REFMAC using the maximum likelihood approach (Murshudov et al., 1997), incorporating bulk solvent corrections, anisotropic Fo versus Fc scaling and TLS refinement, with each subunit defining a TLS group. Random sets of reflections were set aside for cross-validation purposes. Automated solvent building was carried out using ARP/wARP (Perrakis et al., 1999) or Coot (Emsley and Cowtan, 2004). Data collection and refinement statistics are reported in Table I.

The final structures comprise residues His1–Arg207/208 for each of the five subunits in the pentamer. The C-terminal dipeptide, Ala209–Gly210, could be resolved only for two subunits in the tropisetron complex. High temperature factors and weak electron densities are associated with residues Asn15–Met19 (devoid of Pro18–Met19 dipeptide in the anabaseine and 4-OH-DMXBA complexes) and residues Tyr188–Cys191 at the tip of loop C in

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one subunit (4-OH-DMXBA complex). In all structures, most of the N-terminal FLAG epitope and a well-ordered GlcNAc moiety linked to Asn74 are visible. Apart from flexible loop regions, the residue positions in the five subunits within a pentamer and between pentamers are very similar in the four structures. Bound anabaseine could be fully resolved as a cyclic form in two of the five binding sites per pentamer (labelled A and B in Figure 2A) and as an open-chain ammonium ketone form in two other binding sites (labelled C). A PEG molecule, arising from the crystallization liquor, was observed in the fifth binding site. Limited binding site occupancy by anabaseine might arise from the depletion of the high affinity, cyclic form, resulting from conversion (Zolletzsch et al., 1989) to the very low affinity, open-chain ammonium ketone at the pH of crystallization (see Figure 6). In the other three complexes, all five binding sites were fully occupied, consistent with the higher affinity and chemical stability of these compounds compared with anabaseine. The stereochemistry of each structure was analysed using MolProbity (Davis et al., 2007); no residues were found in the disallowed regions of the Ramachandran plot.

Atomic coordinates and structure factors of the A-ACHBP complexes with anabaseine, DMXBA, 4-OH-DMXBA and tropisetron have been deposited with the Protein Data Bank (see Table I for accession codes). Figure 1 was generated using ChemDraw (CambridgeSoft, Cambridge), Figures 2–5 using PyMOL (DeLano, 2002) and Figure 6 using GraphPad Prism 4.0 (GraphPad Software, San Diego).

Crystal packing analysis
For all structures, systematic analysis of the crystal packing contacts within 4.2 Å of residues Glu186–Tyr195 in loop C was carried out using the NCONT program (CCP4). Overall, the residue pair Gin186–His187 along with Ser189 at the base of loop C from one to two subunits within each pentamer establish crystal contacts with a neighbouring pentamer. Regardless of the participation, or a lack thereof, of loop C in crystal contacts between adjacent pentamers, its position remains unchanged, indicating that these contacts have no influence on the position of the loop C tip. Instead, residues within the base of loop C may contribute to the large number of crystal packing geometries documented as seen in the large diversity (>20) of space groups and cell dimensions that have been currently reported for crystals of AChBP.

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Supplementary data
Supplementary data are available at The EMBO Journal Online (http://www.embojournal.org).

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Conflict of interest
The authors declare that they have no conflict of interest.
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